

THE INNOVATORS

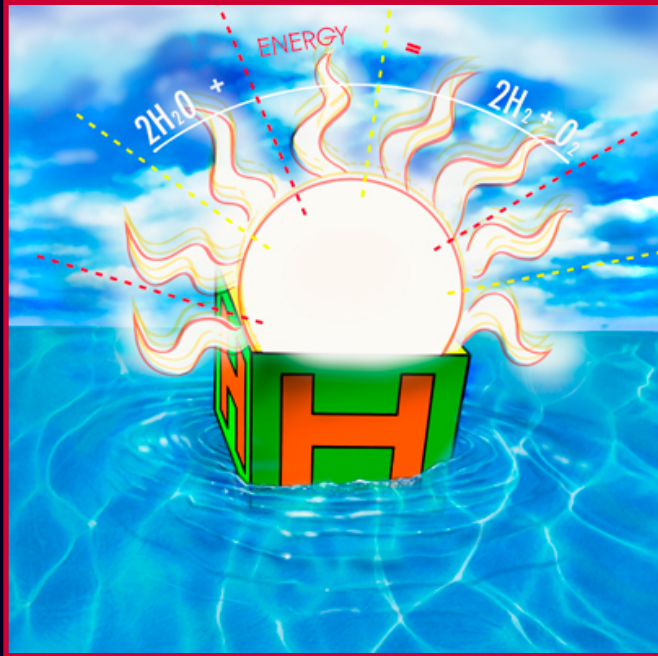
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Cutting-edge Discoveries: Transforming Lives, Fueling the Economy

Fall 2006 Series



The Innovators

Nanotechnology:

The Power to Fuel an Energy Revolution

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School of Mechanical and Materials Engineering
Associate Dean of Research and Graduate Programs
College of Engineering and Architecture

Context

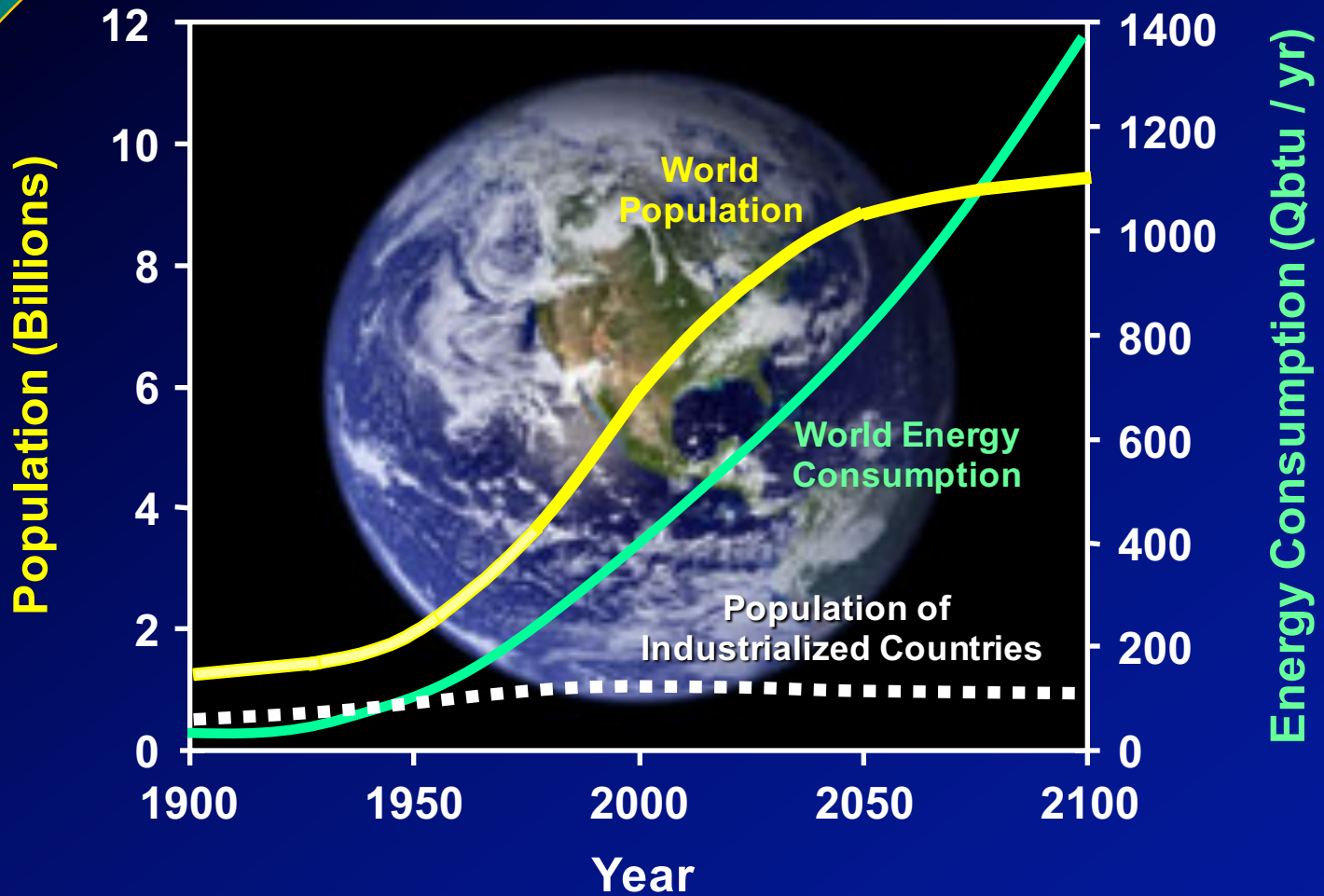
Example 1 - Hydrogen storage

Example 2 - Catalysis

Example 3 - Solar energy conversion

Conclusions

This is the challenge!



Energy Projections: "Global Energy Perspectives" ITASA / WEC
Population Projections: United Nations "Long-Range World
Population Projections: Based on the 1998 Revision"

Inevitable Transition to New Energy Technologies

**19th Century
1800s**

Direct, wood, wind,
water, animals

Steam Engine – Coal
1830-1940



Electric Dynamo –
Coal 1900-1940



Internal Combustion
Engine-Oil
1910-1970



Nuclear
1970-1990



Combined-Cycle
Gas Turbine 1990



**20th Century
1900s**

**21st Century
2000 & beyond**

Fuel Cell
Hydrogen

Direct Electric
Solar

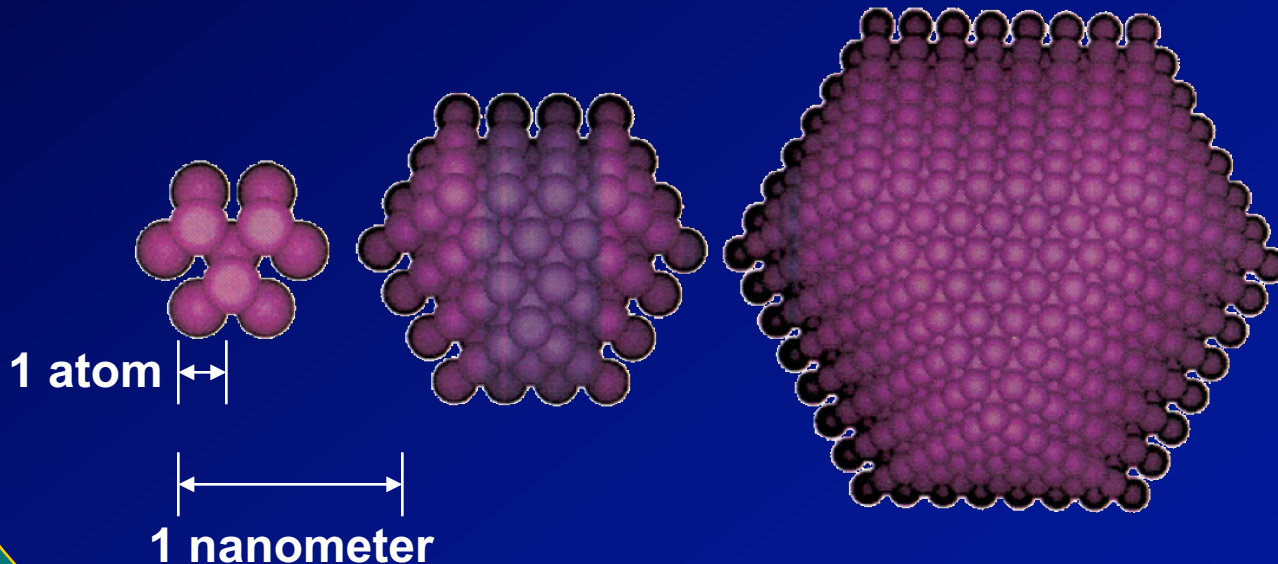
Advanced Biobased
Technologies

Zero Energy
Homes

The Role of Nanotechnology

“Innovations in nanotechnology and other advances in materials science would make it possible to transform our vision of plentiful, low-cost energy into a reality” *Richard E. Smalley*, 1996 Nobel Laureate in Chemistry (June 2005)

“Technology helps and good ideas spread -- these are two laws of nature” Mr. Patel speaking to his wife in *The Life of Pi*, Yann Martel, Harcourt, Orlando (2001)



The Hydrogen Economy

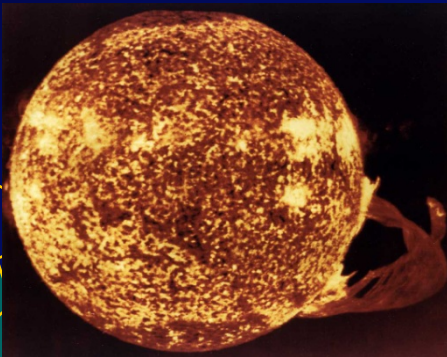
Hydrogen Production → Hydrogen Storage → Hydrogen Conversion

From:
Fossil fuels
Biomass
Water

Either:
Chemically
Physically

To:
Electricity
Heat

Energy Stream



H_2



H_2O



The research needs of the hydrogen economy are quintessentially “nano”

- Catalysis
- Hydrogen storage
- Electrodes for fuel cells

All depend on nanoscale processes and architectures

The Transportation Challenge

Effective Storage is Key

- Enough hydrogen for 300 miles (480 km)
-- about 5-10 kg of useable hydrogen
- Charge/recharge near room temperature
- Quick uptake and release (refueling in 5 minutes)
- None of the current approaches is close to meeting targets



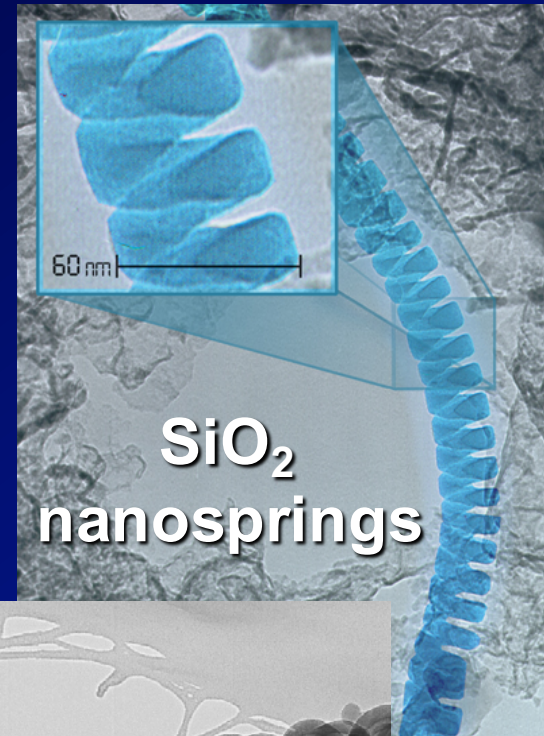
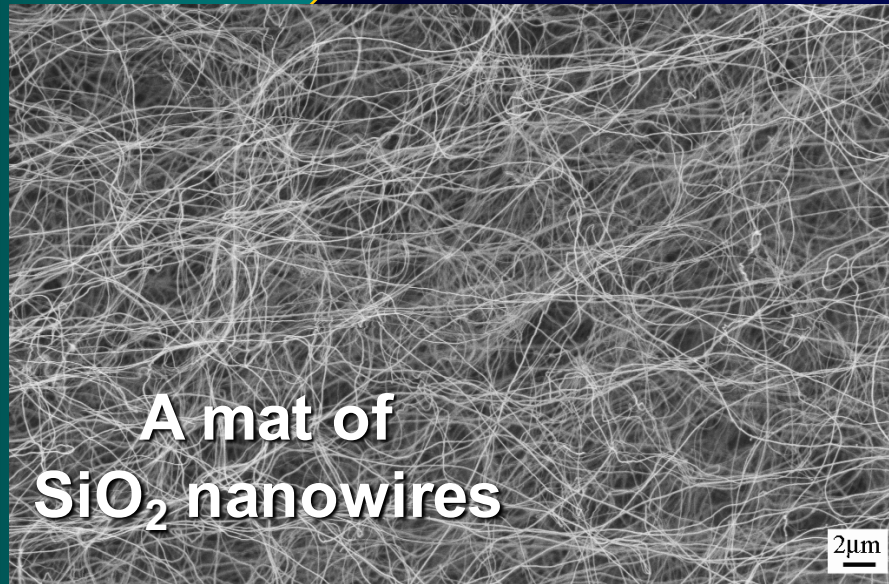
Our Approach

- **Attach molecular hydrogen to the surface of nanomaterials through weak surface-molecular bonds**

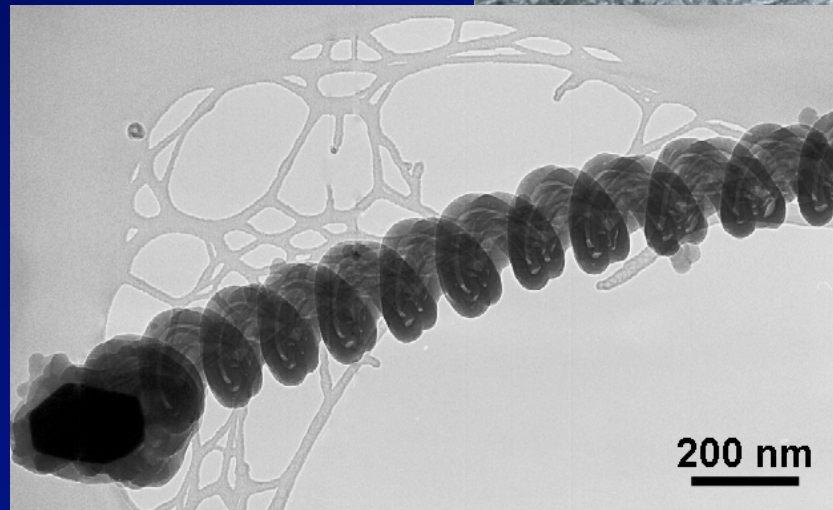
Our Material

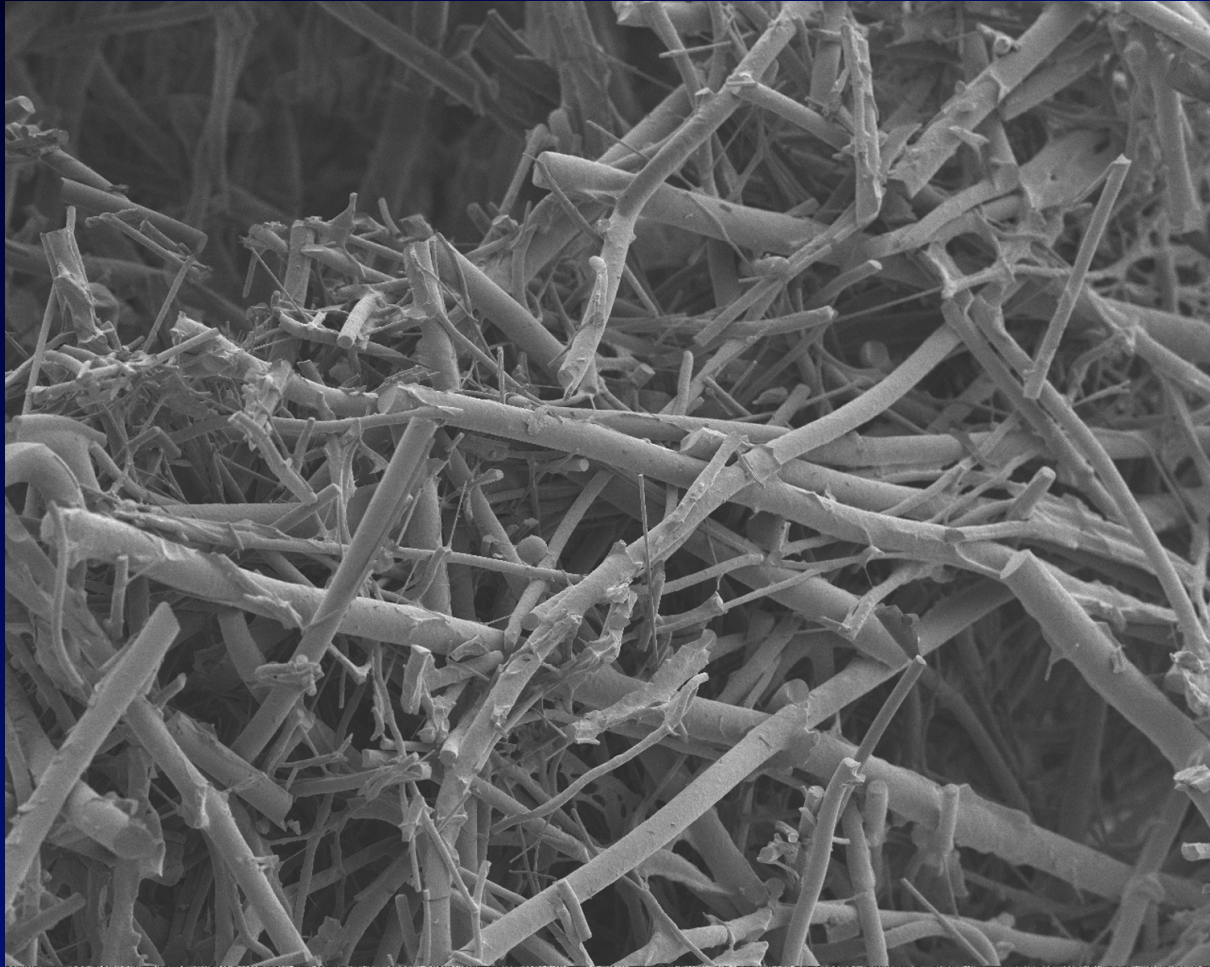
- **One dimensional nanostructured glass “springs”**

1D Silica Nanostructures



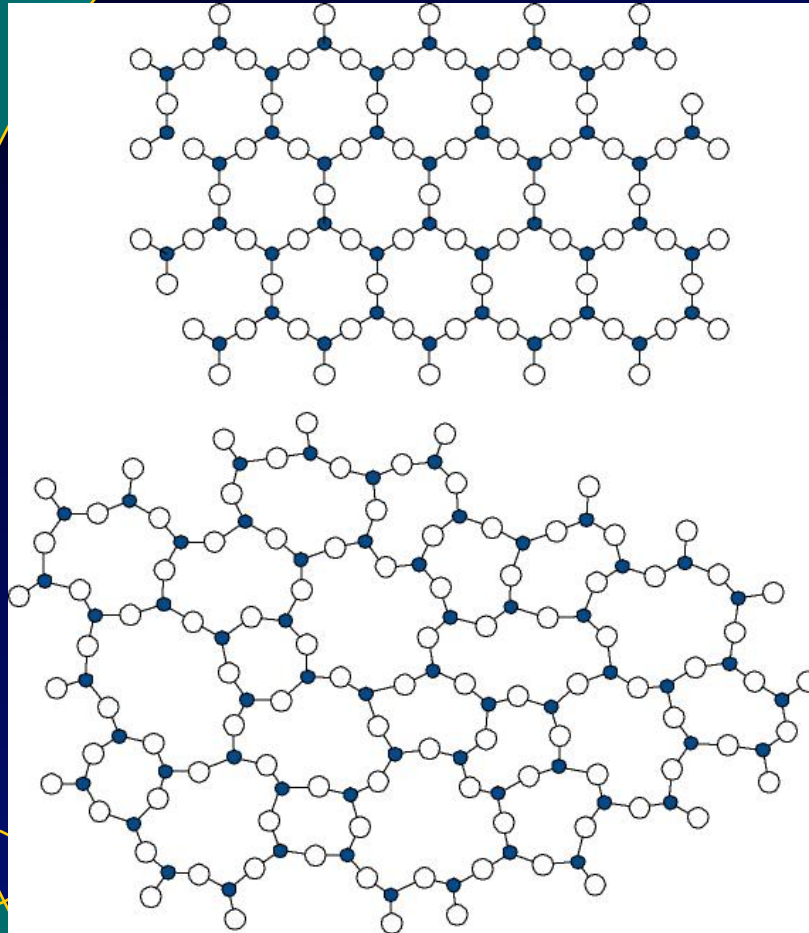
SiO_2
nanocoils



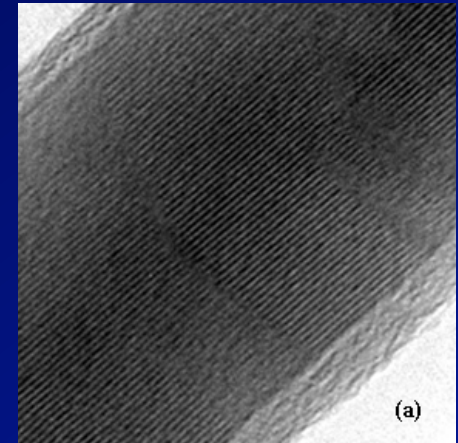


**These are silica glass - this is the surface
of the space shuttle tiles.
200 nanosprings could fit in each fiber.**

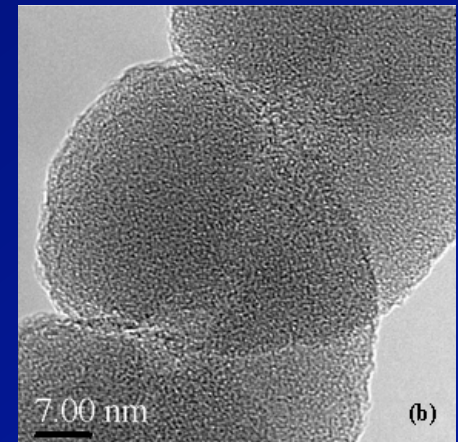
What is a Glass?



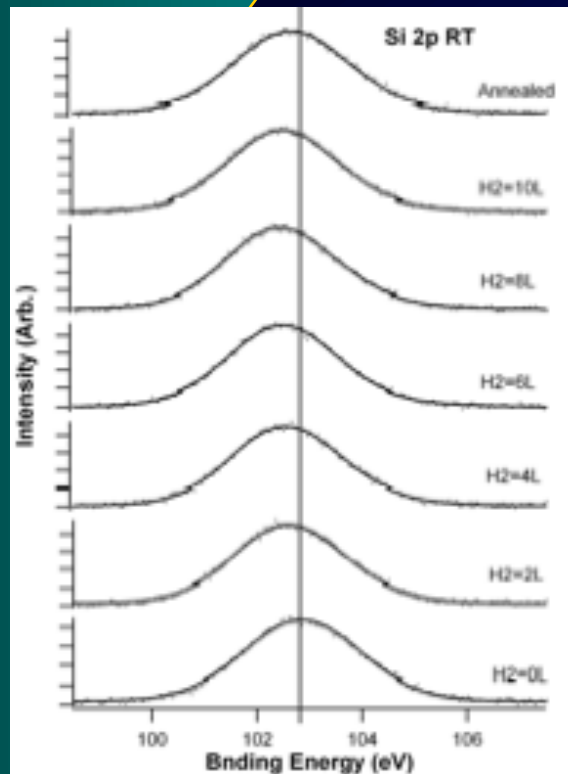
A crystal
Order



A glass
Disorder



How Do We Know H₂ Attaches?



Si 2p XPS spectra

More H₂ in system

- We measure shifts in binding energy
- We form a monolayer (one molecule thick) of H₂ on surface
- Additional layers then form on top (coadsorption) This is unique to this system
- The H₂ bonds only to the silicon
- It goes on at 25°C
- It comes off at 100°C
- This is better than any current alternative
- How much goes on?
More than 5% (gravimetric)
More than 70% (volumetric)

Need to go from the nanoscale
to the macroscale

This can be achieved by forming
the nanosprings on polymers
(plastics), which can be produced
cheaply in complex 3D shapes,
e.g., a honeycomb

Fill the channels with nanosprings



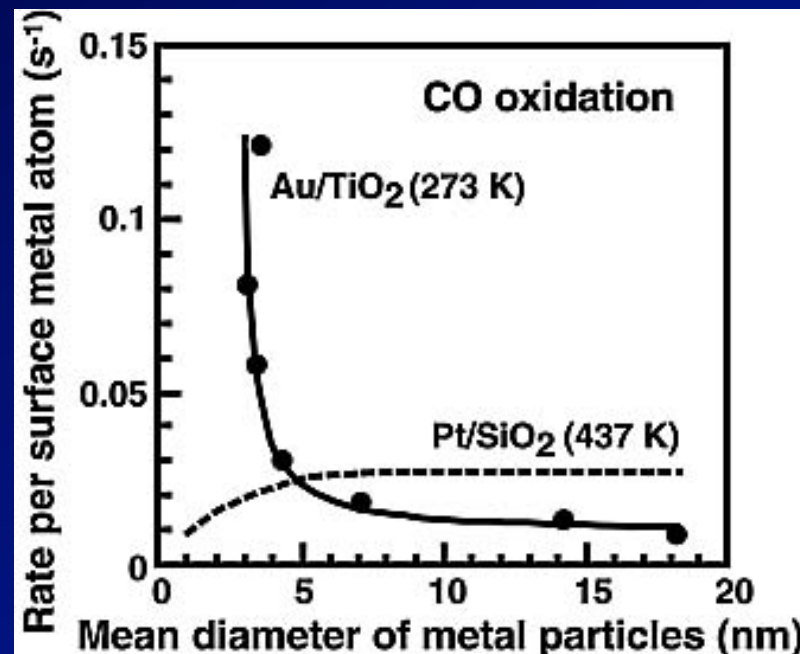
Nanoparticle Catalysts

Catalysts are central
to energy conversion

An example is the water gas
shift reaction for H₂ production



“Water Splitting”



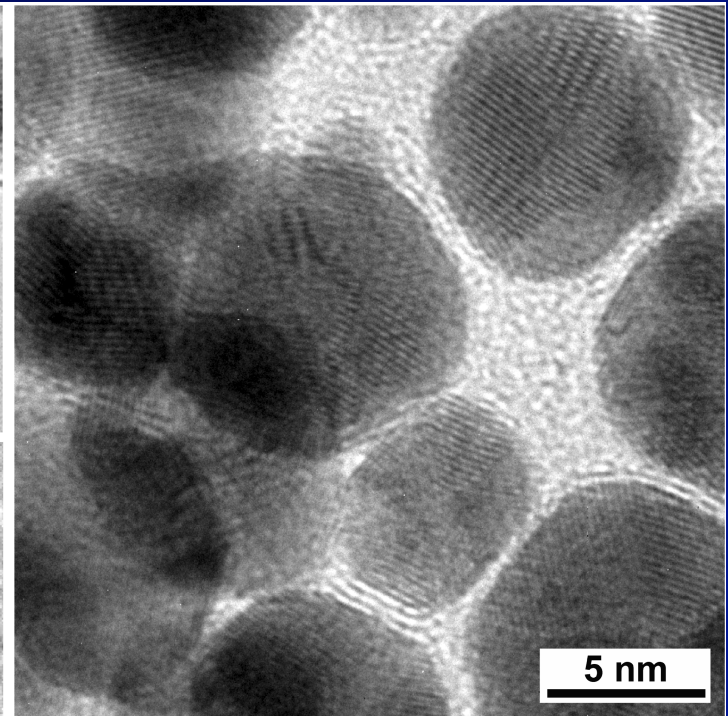
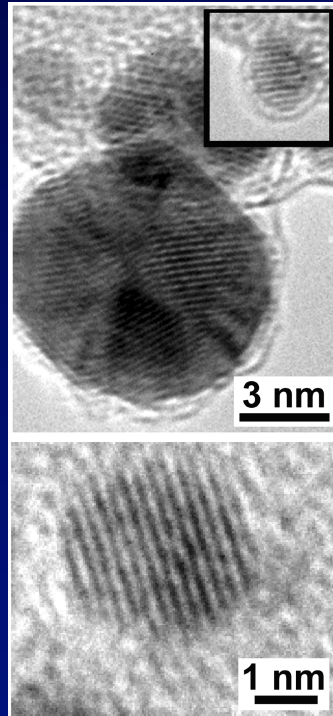
Reaction rate as a function
of nanoparticle size

Gold is only catalytically
active at sizes < 10 nm

Nanoparticle Gold is Particularly Exciting



1350 BCE
Inert



2006
Potent catalyst

Nanoparticle Gold is Not New

‘Purple of Cassius’ after Andreas Cassius

- 1685
- Color is due to small gold particles



Lycurgus Cup

- Rome 4th Century
- Dichroic — color due to colloidal gold and silver

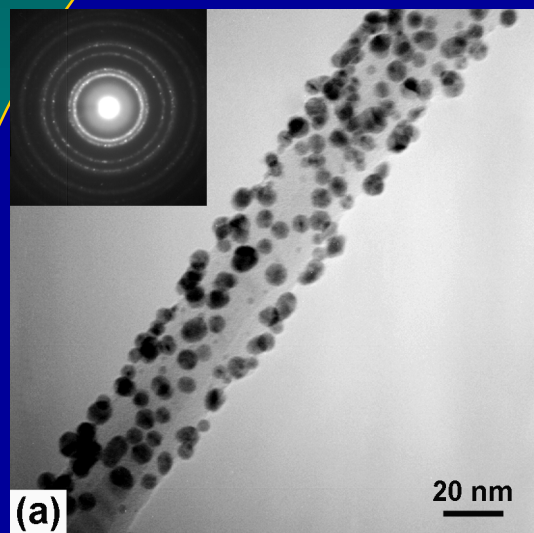


Reflected light

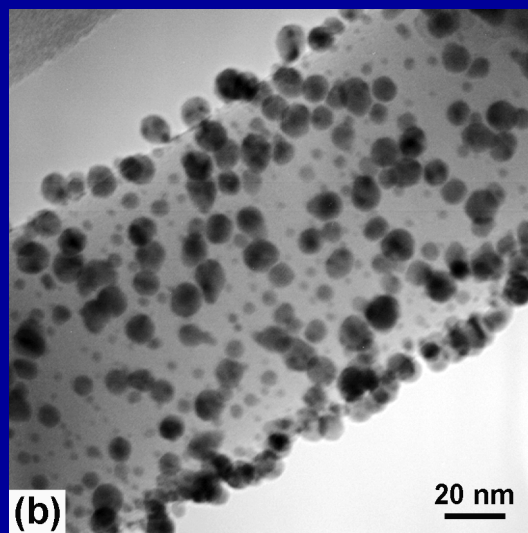
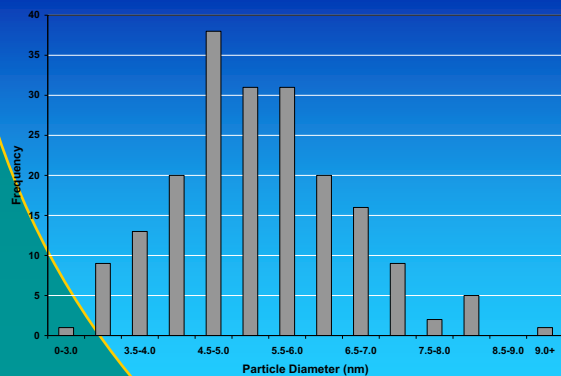


Transmitted light

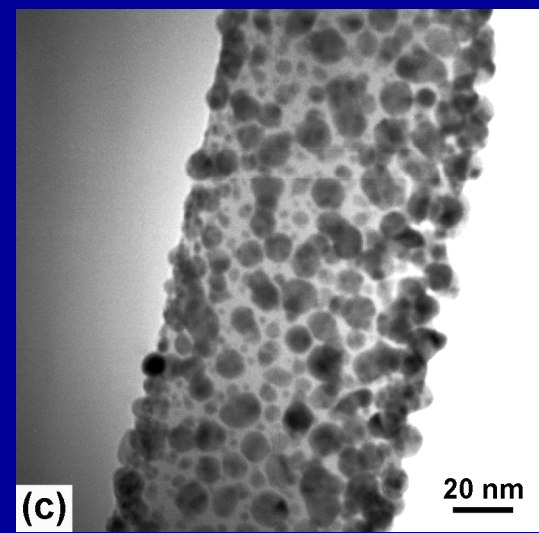
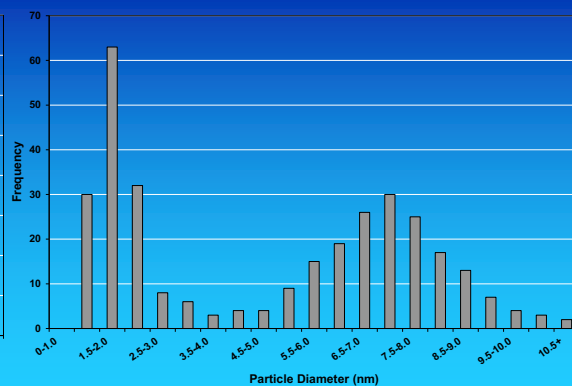
1D Nanostructures as Catalyst Supports



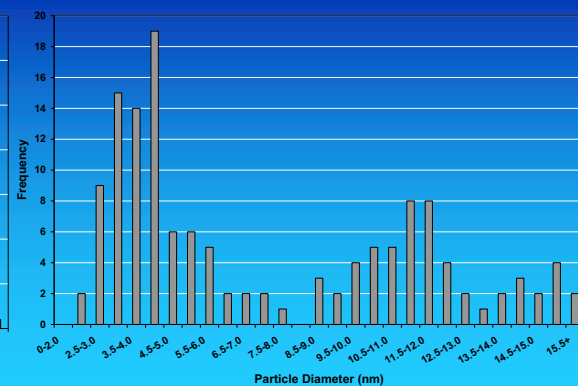
$T = 573\text{ K}$

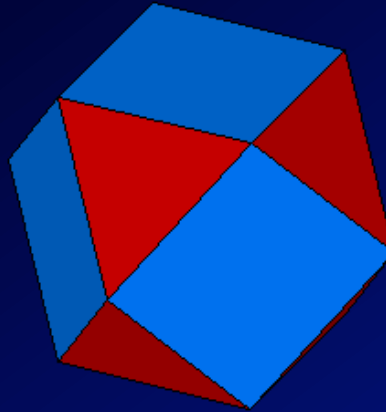


$T = 723\text{ K}$



$T = 873\text{ K}$





Small particles are cubeoctahedra

Significant fraction of atoms occupy surface sites
—not all surface sites are equal

Example:

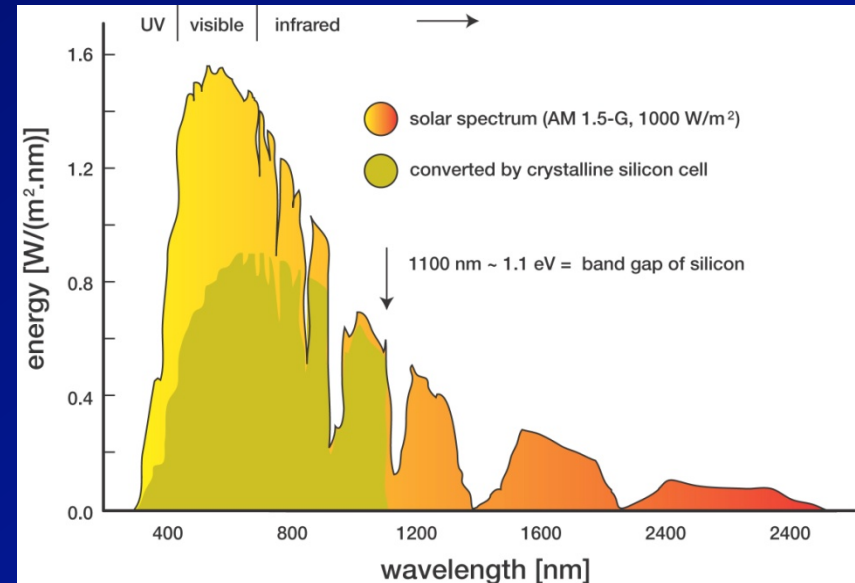
C=O groups preferentially activated on {111} surfaces;
C=C activated at corner and edge sites

For 3 nm cubeoctahedron:

- | | | | |
|----------------|-----|-------------|-----|
| • Corner atoms | 5% | Edge atoms | 25% |
| • {100} faces | 10% | {111} faces | 60% |

Solar Energy

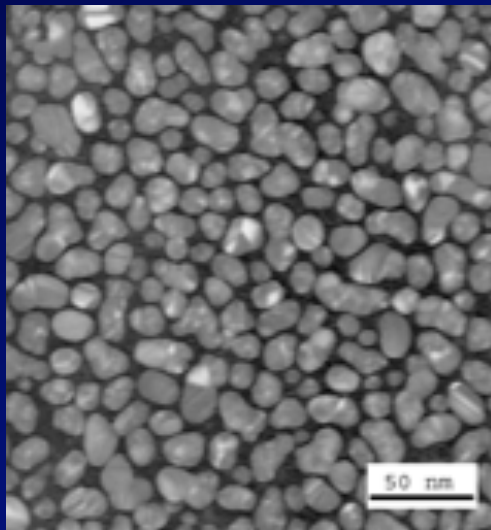
- Sunlight provides by far the largest of all carbon-neutral energy sources
- Solar energy striking the Earth in one hour 4.3×10^{20} J
- Energy consumed on planet in one year 4.1×10^{20} J
- Solar energy provides less than 0.1% of world's electricity



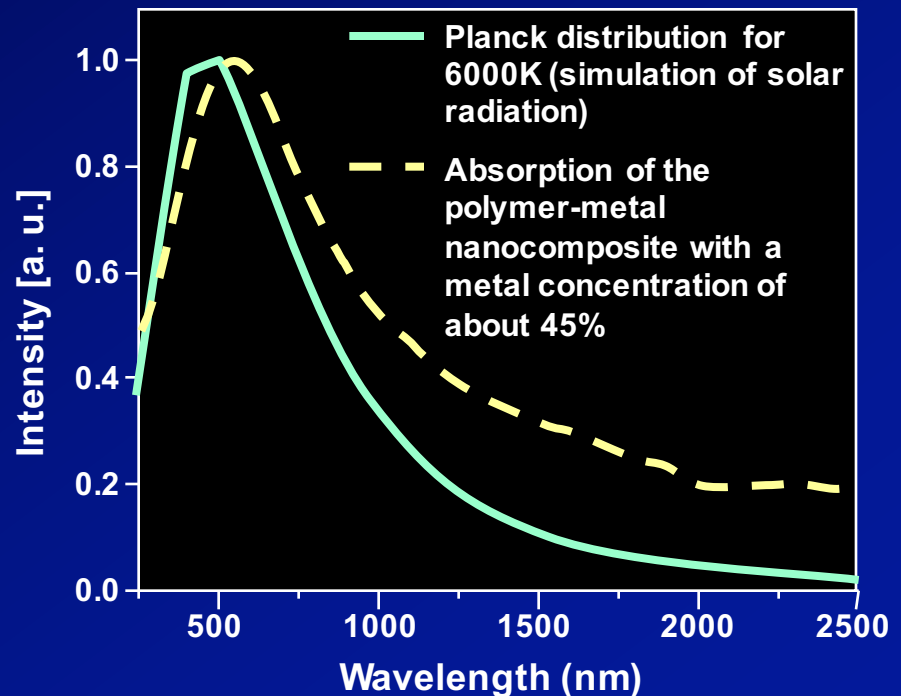
Solar Energy Systems

Nanomaterials may have vital role in improving energy efficiency of solar cells

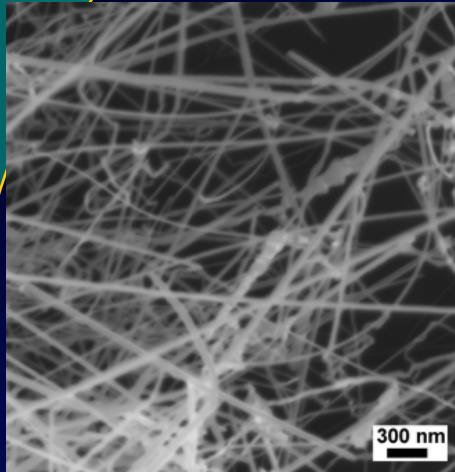
- Efficiency of conventional solar cells limited by absorption range
- Metal nanoparticles have the potential to harvest more of the sun's energy
- Ease of fabrication
- Scale-up



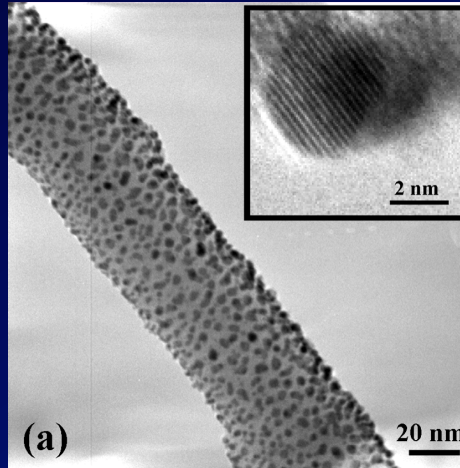
Ag nanoparticles in Teflon AF



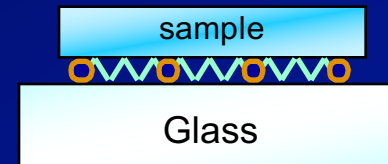
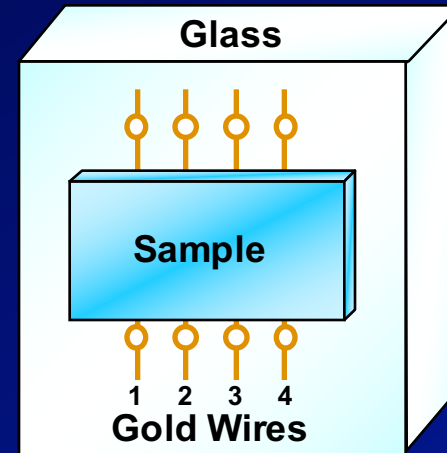
Active Nanosystems for Solar Energy Conversion



Take a mat of
semiconducting
nanowires
(Gallium nitride)



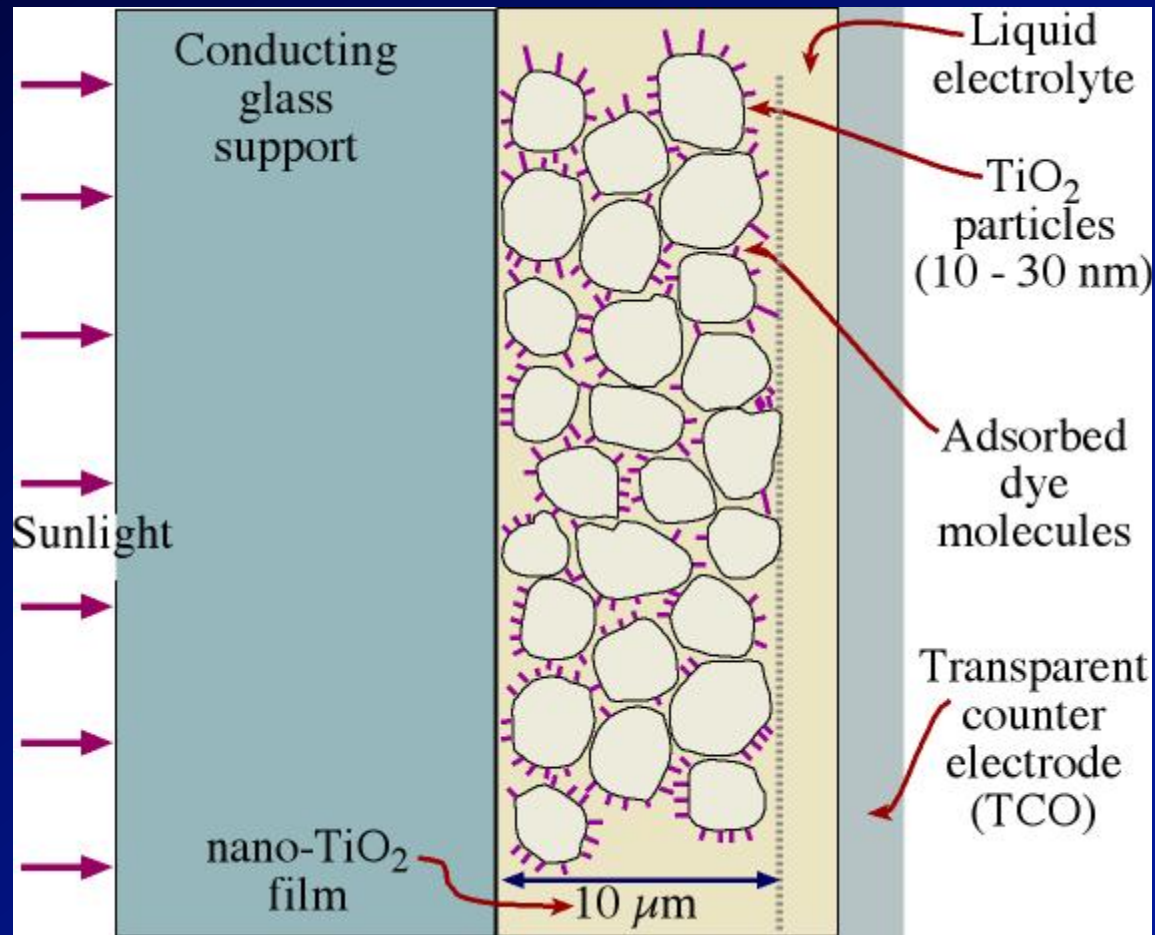
Deposit metal
nanoparticles



Build a device

- Glass
- Sapphire
- Gold Wire
- ~ Nanowires

Other Approaches Using Nanomaterials Dye-sensitized Nanocrystalline Solar Cell



Energy Research — Connecting the Pieces

- **Nanomaterials**
- **“Green” architecture**
- **Bioenergy**
- **Solar cells**
- **Energy harvesting**
- **Energy efficiency**
- **Fuel cells**



Washington State University has strengths in all these areas — our approach is to coordinate these activities with public policy and outreach to address this grandest of the “grand challenges.”

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